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Tyrosine- and tryptophan-coated gold nanoparticles inhibit amyloid aggregation of insulin

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Abstract Here, we have strategically synthesized stable gold (AuNPs^{Tyr}, AuNPs^{Trp}) and silver (AgNPs^{Tyr}) nanoparticles which are surface functionalized with either tyrosine or tryptophan residues and have examined their potential to inhibit amyloid aggregation of insulin. Inhibition of both spontaneous and seed-induced aggregation of insulin was observed in the presence of AuNPs^{Tyr}, AgNPs^{Tyr}, and AuNPs^{Trp} nanoparticles. These nanoparticles also triggered the disassembly of insulin amyloid fibrils. Surface functionalization of amino acids appears to be important for the inhibition effect since isolated tryptophan and tyrosine molecules did not prevent insulin aggregation. Bioinformatics analysis predicts involvement of tyrosine in H-bonding interactions mediated by its C=O, -NH₂, and aromatic moiety. These results offer significant opportunities for developing nanoparticle-based therapeutics against diseases related to protein aggregation.

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Abbreviation

CD Circular dichroism

AuNPs^{Tyr} Tyrosine-coated gold nanoparticles
AuNPs^{Tyr} Tryptophan-coated gold nanoparticles
AgNPs^{Tyr} Tyrosine-coated silver nanoparticles

 $T_{\rm m}$ Transition temperature

Introduction

Making of nanoparticles which are strategically surface coated or functionalized with potential drugs has become one of the most useful tools to target a number of medical complications. Aggregation of proteins into amyloid fibrils is a very important process in biology and a number of pathologies are known to be associated with amyloid aggregation of proteins (Aguzzi and O'Connor 2010; Greenwald and Riek 2010). So far ~35 different proteins are known to be associated with the onset of many neurodegenerative diseases (Greenwald and Riek 2010; Chiti and Dobson 2006). The process of protein aggregation is also a matter of much concern for the storage of protein therapeutic agents (Swift 2002). One of the straightforward approaches to target amyloid-linked diseases and its associated medical complication is to find potential inhibitors against the onset of amyloid aggregation of proteins. Candidates ranging from single amino acids (Kar and Kishore 2007; Shiraki et al. 2002; Ghosh et al. 2009) and natural products (Stefani and Rigacci 2013) to selected peptides (Etienne et al. 2006; Rajasekhar et al. 2015; Viet et al. 2011) have been reported to interfere with the amyloid aggregation of associated proteins. Few investigations



have also looked at the effect of various nanoparticles on amyloid aggregation of proteins (Alvarez et al. 2013; Majzik et al. 2010; Gao et al. 2011). Citrate-capped gold nanoparticles of size greater than ~20 nm are known to delay the growth rate of α -synuclein aggregation (Alvarez et al. 2013). In a recent study, albumin-modified magnetic nanoparticles were found to promote the depolymerization of insulin amyloid fibrils (Siposova et al. 2012). Successful synthesis of potential gold nanoparticles using various amino acids has been reported recently (Maruyama et al. 2014; Daima et al. 2013). Gold nanoparticles coated with different amino acids have also shown substantial effect on serum albumin absorption and cytotoxicity (Cai et al. 2014).

Therefore, the foundation of the current research was laid on our belief that control of protein aggregation can be efficiently achieved by the presence of amino acid surface corona on the surface of gold and silver nanoparticles, and it may provide additional opportunities to control various physicochemical properties of nanoparticles in controlled manner. For example, amino acids are zwitterionic, and therefore, surface charge of amino acid-coated nanoparticles can be tailored by varying the solution pH. Moreover, the presence of amino acid coating on metal nanoparticles may provide biological identity to nanoparticles.

It is believed that hydrophobic residues including aromatic amino acids play a critical role during amyloid aggregation of proteins (Chiti and Dobson 2009; Fandrich et al. 2001). Since exposed hydrophobic residues are vital to the onset of protein aggregation, adapting a strategy to interfere with the intermolecular hydrophobic interactions may possibly provide an effective approach to inhibit such aggregation process. In accordance with this hypothesis, we have attempted to make metallic nanoparticles surface functionalized with selected hydrophobic residues (tyrosine and tryptophan), and have studied their inhibition effect on amyloid aggregation of a globular protein, insulin.

Here, we demonstrate inhibition of insulin amyloid aggregation using tyrosine-coated gold nanoparticles (AuNPs^{Tyr}), tyrosine-coated silver nanoparticles (AgNPs-^{Tyr}), and tryptophan-coated gold nanoparticles (AuNPs^{Trp}). Strong inhibition of both spontaneous and seed-induced aggregation of insulin was observed in the presence of these nanoparticles. We have also carried out bioinformatics analysis and molecular docking studies to understand possible molecular interactions between tyrosine and insulin. Ultimately, this work provides a new approach to target aggregation of proteins by stable nanoparticles coated with aromatic residues.

Materials and methods

Reagents

Tetrachloroauric acid (HAuCl₄), AgNO₃, L-tryptophan, L-tyrosine, potassium hydroxide (KOH), Thioflavin T, acetic acid, and sodium hydroxide (NaOH) were purchased from Sigma–Aldrich. Deionized MilliQ water was used to prepare various gold and silver nanoparticles. Lyophilized powder of bovine insulin, phosphate buffer saline (PBS), and L-phenylalanine were obtained from HIMEDIA (India).

Synthesis of amino acid-conjugated gold and silver nanoparticles

To synthesize tryptophan and tyrosine amino acid-conjugated gold and silver nanoparticles, 100 ml aqueous solutions of 1 mM KOH containing 0.5 mM tryptophan and 0.1 mM tyrosine were separately allowed to boil under vigorous stirring experimental condition (Selvakannan et al. 2013). In the stirring boiling solutions, [AuCl₄] or Ag⁺ ions were added to obtain tryptophan or tyrosine amino acid-conjugated gold and silver nanoparticles, respectively. The total concentrations of both the [AuCl₄] or Ag⁺ metal ions were kept constant at 0.2 mM in all the reactions. Through this process two different gold (AuNPs^{Tyr}, AuNPs^{Trp}) and silver nanoparticles (AgNPs^{Tyr}) solutions were obtained. All these nanoparticle solutions thus obtained were cleaned from the unreduced metal ions and unbound amino acids by dialyzing these solutions against deionized water using 12 kDa dialysis membranes for 3 h followed by overnight dialysis. Even after dialysis, all these gold and silver nanoparticles' solutions remained stable without showing any sign of aggregation, indicating that these nanoparticles were strongly capped with amino acids. All the solutions of tryptophan and tyrosine amino acid-conjugated gold and silver nanoparticles were found to be stable under standard laboratory storage conditions at room temperature for more than 6 months, and used as such for characterization and biological studies.

Fluorescence measurements

A Perkin-Elmer LS 55 fluorescence spectrometer was used to monitor the amyloid aggregation of proteins by Thioflavin T binding assays where Thioflavin T was excited at 440 nm with a slit width of 5 nm and the emission was observed at 490 nm (Morozova-Roche et al. 2000).



UV-Visible spectroscopy

UV-Visible spectrophotometer (Varian Cary-4000 and Shimadzu UV-1800) were used to measure concentrations of protein samples (Dubey et al. 2014) and to obtain UV-Vis spectra of tryptophan-and tyrosine-coated gold and silver nanoparticles.

FTIR spectroscopic analysis

FTIR spectra of tryptophan- and tyrosine-coated gold and silver nanoparticles were recorded in DRS mode using Perkin-Elmer D100 spectrophotometer with a resolution of 4 cm⁻¹.

Amyloid aggregation of insulin

Amyloid aggregation of insulin was studied by incubating the monomer samples (~40 μ M) in 10 mM PBS at ~70 °C in the presence and in the absence of AuNPs^{Tyr}, AuNPs^{Trp}, and AgNPs^{Tyr} nanoparticles (Mark et al. 2004; Chatani et al. 2014). Small aliquots of these aggregating samples were taken out at regular intervals and their Thioflavin T fluorescence intensities were recorded. For conducting seed-induced amyloid aggregation reactions, pre-formed amyloid fibrils (~15 % weight/weight) were used as seeds. For fibril disassembly experiments, a suspension of mature insulin amyloid fibrils was prepared and Thioflavin T signal of the sample was recorded at different time intervals in the presence and absence of AuNPs^{Tyr} (molar ratio of insulin fibrils to AuNPs^{Tyr} was maintained at 1:1).

Circular dichroism spectroscopy

For the CD experiments, JASCO CD spectrometer (model J-815-150 L) with attached Peltier temperature controller was used. The path length of the cell was 2 mm. The structural changes in the protein sample was observed in the presence and absence of AuNPs^{Tyr} and AuNPs^{Trp}, by monitoring the CD scans at different time points at 70 °C. Thermal unfolding of insulin was performed by continuously recording the spectra while sample was heated from 25 to 90 °C at a scan rate of 2 °C min⁻¹.

Transmission electron microscopy

Transmission electron microscope (HR-TEM JEOL JEM-2100 and JEOL 1010) was used to examine mature amyloid fibrils and AuNPs^{Tyr}, AuNPs^{Trp}, and AgNPs^{Tyr} nanoparticles, respectively. Mature amyloid fibril samples were spotted on a carbon-coated grid for 2 min and the samples were then washed with water. The samples were then stained with 1 % (w/v) aqueous uranyl acetate solution for

~2 min followed by another washing step. Air-dried grids were then examined.

Molecular docking and bioinformatics analysis

The structure of ligand (L-tyrosine) was obtained from PubChem (CID 6057) (Bolton et al. 2008) and was prepared using 'prepare ligand wizard' of Discovery studio 4.0 (DS4). X-ray crystal structure of insulin (PDB ID: 4I5Z) (Pechkova et al. 2014) was also prepared through 'prepare protein wizard' of DS4, and pre-processing and protonation were carried out using CHARMm force fields (Brooks et al. 2009). The ligand and protein were then docked using a blind approach (no active site defined) with 100 conformations to choose and 100 orientations to refine by following CDocker protocol (Wu et al. 2003). The complex was then typed with CHARMm force field, and was subjected to molecular dynamics simulation for 5000 picoseconds under NTP (normal temperature and pressure) and distance-dependent dielectrics (implicit solvent model as a crude approximation of polar solvent). The simulations were executed in a Dell precision T5610 workstation with 16 processors and 32 GBs of RAM. Visualizations were performed using DS4. For analysis of crystal structure of bovine insulin multimers, we obtained four multimeric bovine insulin crystal structures from PDB sources (two hetero 12-mers; PDB IDs: 2A3G and 2INS, and two hetero 6-mers; PDB IDs: 2ZP6, 4IDW) (Smith et al. 1982, 2005; Margiolaki et al. 2013). The interface of insulin structures was analyzed to identify the key amino acids involved in multimer formation using DS4. The structures were cleaned by removing water and heteroatoms leaving behind nascent molecules.

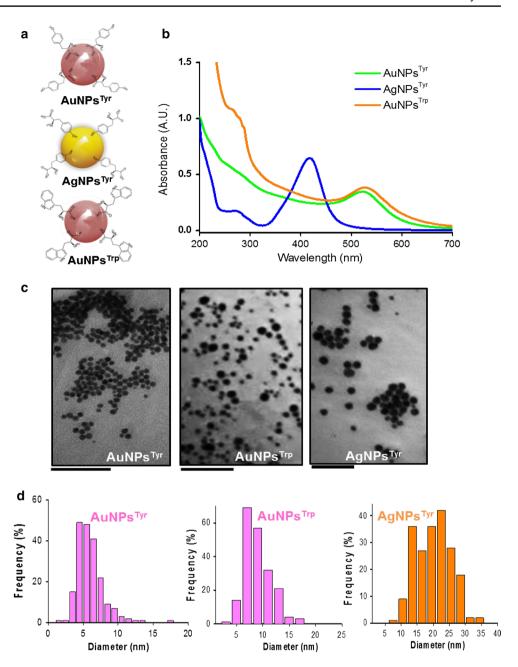
Results

Characterization of tyrosine- and tryptophan-coated nanoparticles

Samples of gold nanoparticles coated with tyrosine (AuNPs^{Tyr}) and tryptophan (AuNPs^{Trp}) and samples of tyrosine-coated silver nanoparticles (AgNPs^{Tyr}) were synthesized by following the established protocol (Selvakannan et al. 2013) (see "Synthesis of amino acid-conjugated gold and silver nanoparticles"). SPR absorption bands are considered as characteristic signatures for the existence of stable metal nanoparticles and therefore the formation of nanoparticles can be confirmed using UV–Visible absorption spectroscopy. UV–visible spectra of the obtained nanoparticles are shown in Fig. 1b. AuNPs^{Tyr} and AgNPs-Tyr nanoparticles showed typical SPR bands at ~523 and ~417 nm, respectively, wherein the SPR absorbance peak



Fig. 1 Characteristic features of tyrosine- and tryptophancoated nanoparticles. a Schematic representation of tyrosine (AuNPs^{Tyr} and AgNPs^{Tyr})- and tryptophan (AuNPs^{Trp})-coated nanoparticles. The orientation of the attachment of tyrosine residues is reversed in the silver nanoparticles (AgNPs^{Tyr}). **b** UV-Visible spectra of the nanoparticle samples: (olive curve) AuNPs^{Tyr}; (orange curve) AuNPs^{Trp}; and (violet curve) AgNPsTyr. c Transmission electron microscopic images of AuNPs^{Tyr}, AuNPs^{Trp}, and AgNPs^{Tyr}. The scale bar represents 100 nm. The average diameter of the nanoparticles was observed to fall within ~15-30 nm. d Particles size histograms corresponding to tyrosine- and tryptophan-coated AuNPs^{Tyr}, AuNPs^{Trp}, and AgNPs^{Tyr} nanoparticles. This analysis was done by using ImageJ software (color figure online)



of AuNPs^{Trp} sample was observed at ~525 nm. The SPR absorption curves of AuNPs^{Tyr}, AuNPs^{Trp}, and AgNPs^{Tyr} nanoparticles look very similar to the reported values for similar nanoparticles. Transmission electron micrographs of AuNPs^{Tyr}, AuNPs^{Trp}, and AgNPs^{Tyr} samples are shown in Fig. 1c. All these nanoparticles appeared spherical displaying a uniform diameter value of ~10–30 nm by analysis of TEM images using ImageJ tools. These amino acid-capped nanoparticles were also observed to be very stable and we did not see any sign of aggregation even after 6 months of storage time.

FTIR spectroscopic analysis of gold and silver nanoparticles synthesized by tryptophan and tyrosine was carried out to understand the surface chemistry of different nanoparticles and to provide direct evidence of amino acids interaction with the nanoparticles. FTIR data are presented in supplementary Figure S14 which shows vibrational frequencies in tryptophan- and tyrosine-coated nanoparticles. The shift of the carbonyl stretching frequency of tryptophan from *1666 cm⁻¹ (magenta curve, Figure S14) to 1707 cm⁻¹ was observed for AuNPs^{Trp} (blue curve, Figure S14). Tyrosine has a carbonyl stretching frequency around *1610 cm⁻¹ (green curve, Figure S14) that shifted to 1740 cm⁻¹ in case of AuNPs^{Tyr} and AgNPs^{Tyr} (red and black curves Figure S14) after the synthesis of nanoparticles.



Inhibition of insulin amyloid aggregation

Amyloid aggregation of insulin in PBS was studied at an elevated temperature, ~70 °C (close to the $T_{\rm m}$ of insulin) by monitoring the rise in the Thioflavin T signal at different time points (see "Materials and methods"). Soluble insulin monomers at ~40 μ M showed a characteristic aggregation curve (Fig. 2, *filled circle*) composing of a distinct lag phase, a growth phase, and a saturation phase. Final aggregates of insulin showed typical amyloid morphology (Fig. 3c) as seen in the previous studies (Mark et al. 2004). The inhibition effect of amino acid-coated nanoparticles was examined at three different molar ratio values (of insulin:inhibitor) as mentioned in Fig. 2. Both AuNPs^{Tyr} and AuNPs^{Trp} showed dose-dependent inhibition of insulin amyloid aggregation (Fig. 2a, b). Control experiments of insulin aggregation in the presence of the solvents and

solutes used for nanoparticle preparation did not show any inhibition effect (data not shown).

To confirm the inhibition effect of coated tyrosine and tryptophan residues, as a next step, we tested the effect of isolated aromatic residues (phenylalanine, tyrosine, and tryptophan) on amyloid aggregation of insulin. The data as shown in Fig. 2d clearly indicate that the control insulin sample and the amino acid-treated sample follow similar kinetics during amyloid aggregation. This result confirms that tyrosine and tryptophan molecules are effective in inhibiting the amyloid aggregation only when they are attached to the surface of the nanoparticles. The inhibition effect of AuNPs^{Tyr} was found to be slightly more suppressing than the effect observed for AuNPs^{Trp} at similar concentration (Fig. 2a, filled triangle, b, inverted filled triangle).

Next, we performed experiments to know the effect of these nanoparticles on seed-induced aggregation of insulin.

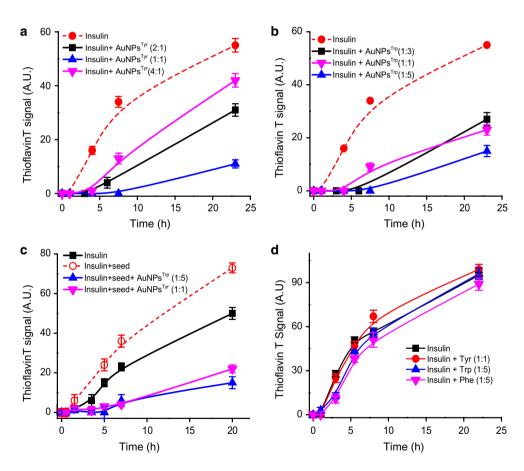


Fig. 2 Inhibition of insulin amyloid aggregation by tyrosine- and tryptophan-coated nanoparticles. **a** Effect of AuNPs^{Tyr} on aggregation of ~40 μM insulin sample at different values of molar ratio: (filled circle) insulin; (inverted filled triangle) insulin + AuNPs^{Tyr} at 4:1; (filled square) insulin + AuNPs^{Tyr} at 2:1; (filled triangle) insulin + AuNPs^{Tyr} at 1:1. **b** Effect of AuNPs^{Typ} on spontaneous aggregation of ~40 μM insulin at different molar ratios: (filled circle) insulin only; (inverted filled triangle) insulin + AuNPs^{Typ} at 1:1 molar ratio; (filled square) insulin + AuNPs^{Typ} at 1:3 molar ratio; (filled triangle)

gle) insulin + AuNPs^{Tyr} at 1:5 molar ratio. **c** Effect of AuNPs^{Tyr} and AuNPs^{Trp} on seed-induced aggregation of insulin (filled square) insulin only; (open circle) insulin + 15 % (w/w) seed; (inverted filled triangle) insulin + seed + AuNPs^{Tyr} at 1:1 molar ratio; (filled triangle) insulin + seed + AuNPs^{Trp} at 1:5 molar ratio. **d** Effect of isolated tyrosine, tryptophan and phenylalanine residues on aggregation of insulin: (filled square) insulin only; (filled circle) insulin + Tyr at 1:1; (filled triangle) insulin + Trp at 1:5; (inverted filled triangle) insulin + Phe at 1:5 molar ratio (color figure online)



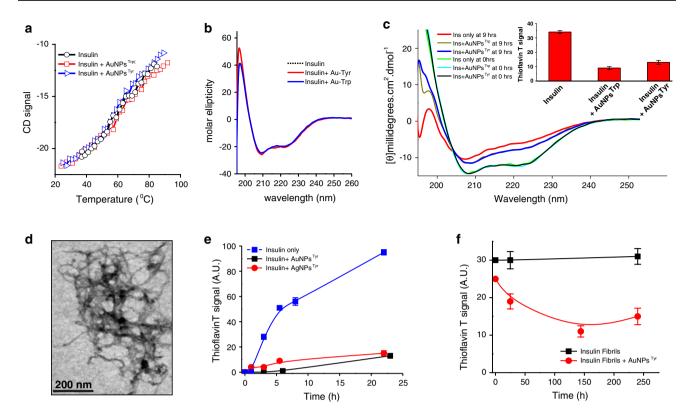


Fig. 3 a CD spectra of insulin in the presence and absence of Trp- and Tyr-coated nano particles: (dotted curve) insulin only; (red curve) insulin + AuNPs^{Tyr} at 1:1 molar ratio; (violet curve) insulin + AuNPs^{Trp} at 1:5 molar ratio. **b** Thermal unfolding of insulin by circular dichroism by monitoring change of CD signal at ~222 nm: (open circle) insulin only; (open square) insulin + AuNPs^{Tyr} at 1:1 ratio; (right headed open triangle) insulin + AuNPs^{Trp} at 1:5 molar ratio. **c** CD spectra of insulin undergoing aggregation in the presence and absence of nanoparticles: insulin only at 0 h; (olive curve) and 9 h (red curve); insulin + AuNPs^{Tyr} at 1:1 molar ratio at 0 h (black

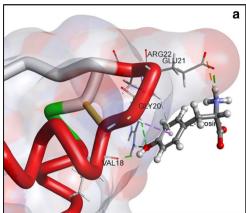
curve) and at 9 h (violet curve); insulin + AuNPs^{Trp} at 1:1 molar ratio at 0 h (blue curve) and at 9 h (black curve). Inset shows ThT signal of the CD samples at 9 h. **d** TEM image of insulin mature amyloid fibrils. **e** Comparison of the effect of AuNPs^{Tyr} and AgNPs^{Tyr} on spontaneous aggregation of ~43 μM insulin: (filled square) insulin only; (filled circle) insulin + AgNPs^{Tyr} at 1:1 molar ratio; (filled square) insulin + AuNPs^{Tyr} at 1:1 molar ratio of insulin: (filled square) insulin fibrils; (filled circle) insulin fibrils + AuNPs^{Tyr} at 1:1 molar ratio of insulin: (filled circle) insulin fibrils + AuNPs^{Tyr} at 1:1 molar ratio of insulin: (filled circle) insulin fibrils + AuNPs^{Tyr} at 1:1 molar ratio of insulin: (folled circle)

In the presence of seed (~15 % w/w), we observed a rapid rise in the Thioflavin T signal without a lag phase. However, inhibition of seed-induced aggregation was observed in the presence of AuNPs^{Tyr} (Fig. 2c, inverted filled triangle) and AuNPs^{Trp} (Fig. 2c, filled triangle) nanoparticles. It appears that these nanoparticles have the ability to interfere with the recruitment of the monomers by mature fibrils which in turn suppresses the seeded elongation during aggregation process. To further clarify the interaction between the AuNPs^{Tyr} and mature fibrils, we examined whether the nanoparticles have the ability to promote dissociation of amyloid fibrils. To conduct disassembly experiments, AuNPs^{Tyr} sample was added to a suspension of mature insulin fibrils, and Thioflavin T signal of the sample was monitored at different time intervals. The obtained data as shown in Fig. 3f clearly indicate that AuNPs^{Tyr} promote disassembly of insulin amyloid fibrils. After ~240 h of incubation, we observed almost ~50 % decrease of the initial Thioflavin T signal (Fig. 3f, filled cirlce) which suggests a substantial amount of fibril dissociation in the presence of AuNPs^{Tyr}. In a previous study, it was reported that albumin modified magnetic fluid causes the depolymerization of insulin amyloid fibrils (Siposova et al. 2012).

Effect on conformational properties of insulin

Since both AuNPs^{Tyr} and AuNPs^{Trp} showed their potential either to inhibit or to interfere with amyloid aggregation of insulin, we used circular dichroism (CD) measurements to understand whether these nanoparticles have any effect on the conformational properties of insulin. First, we examined the CD spectra of the insulin sample in the presence and absence of nanoparticles. The molecular conformation of insulin remained unchanged in the presence of nanoparticles (Fig. 3b). Next, we performed thermal unfolding of insulin in PBS by monitoring the change in the CD signal at \sim 222 nm. The thermal unfolding curves showed no significant change in the $T_{\rm m}$ values of insulin samples in the





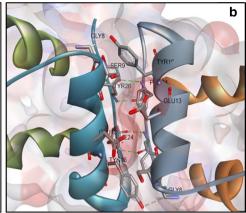


Fig. 4 a Molecular docking studies of a tyrosine molecule with insulin. The complex shows seven interactions comprising of four hydrogen bonds (B:ARG22:HE—L-tyrosine:O14, B:ARG22:HH21—L-tyrosine:O14, B:GLU21:OE1—L-tyrosine:H21, and B:VAL18:O—L-tyrosine:H24), two pi interactions (B:GLU21:HN—L-tyrosine, B:GLY20:HA1—L-tyrosine), and one electrostatic bond

(B:GLU21:OE2—L-tyrosine:N13). **b** Analysis of a crystal structure of insulin dimer to visualize the binding partners of tyrosine residue of one insulin molecule with different functional groups of another insulin molecule at the interface region. Discovery Studio 4.0 was used for the docking studies

presence of these nanoparticles (Fig. 3a, open square, open circle, and right-headed open triangle).

To understand the structural changes during inhibition of aggregation, we also monitored the CD signal of aggregating insulin samples in the presence of and absence of both AuNPs^{Tyr} and AuNPs^{Trp}. Figure 3c compares the CD spectra of both inhibited and uninhibited reactions measured at 0 and 9 h. Data clearly indicate that the process of conversion of native insulin molecules into β -structured species is delayed in the presence of nanoparticles (Fig. 3c, blue curve and olive curve). Thioflavin T readings of the same samples are almost consistent with the observed CD data (inset of Fig. 3c).

Molecular docking of tyrosine with insulin

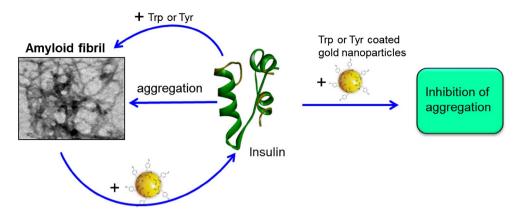
To further understand the observed inhibition effect, we carried out molecular docking studies on insulin-tyrosine interactions. Blind docking studies (see "Materials and methods") predicted a viable tyrosine-insulin interaction at a site within insulin's B-chain which spans from Val18-Arg22. The value of CDocker energy for the best pose (as shown in Fig. 4a) was -29.02 kcal mol⁻¹ and its corresponding interaction energy was observed to be $-22.95 \text{ kcal mol}^{-1}$. Such energy values suggest the formation of a stable protein-ligand complex. Obtained tyrosine-insulin complex as shown in Fig. 4a displays seven interactions composing of four hydrogen bonds (B:ARG22:HE-L-tyrosine:O14, B:ARG22:HH21-L-tyrosine:O14, B:GLU21:OE1—L-tyrosine:H21, and B:VAL18:O—L-tyrosine:H24), two pi interactions (B:GLU21:HN—L-tyrosine, B:GLY20:HA1—L-tyrosine), and one electrostatic bond (B:GLU21:OE2—L-tyrosine:N13).

The stability of insulin-tyrosine complex was further ascertained using molecular dynamics study. Both root-mean square deviation and total energy of the complex indicated the formation of a stable conformation after 5000 ps (Figure S4, supplementary information). These data suggest that tyrosine's docking site spans from Val18-Arg22 region of insulin's chain B and binding of tyrosine to that region of insulin may perhaps result in a stable protein-ligand complex. We further carried out the docking studies on insulin-alanine and insulin-tryptophan interactions. The obtained data (Figure S10, S11, S12, and S13 in the supplementary information) suggest that the number of hydrogen bonds formed between ligand and the protein is highest when the ligand is tyrosine and is lowest when the ligand is alanine. Further the molecular dynamics studies of tyrosine-insulin complex formation predicts a more favorable complex in terms of energy and RMSD than that of insulin-alanine and insulin-tryptophan interactions (Figure S12).

To further clarify the possible binding partners of a tyrosine residue when it interacts with insulin, we looked at the available crystal structures of insulin dimers (PDB IDs: 2A3G, 2INS, 2ZP6, and 4IDW) (Smith et al. 1982, 2005; Margiolaki et al. 2013). We focused on the interface of insulin–insulin dimers and analyzed the involvement of tyrosine residues in intermolecular interactions. Tyr16–Tyr26 region of chain B of insulin was found to be mediating important H-bonding interactions with average hydrogen bond length of 3.055 Å. Further data analysis (Figure S5–S8, supplementary information) clearly indicates that tyrosine residues are playing a key role during insulininsulin dimer formation (Figure S8, S9, supplementary information).



Fig. 5 Schematic representation of inhibition of insulin amyloid formation and promotion of amyloid dissociation in the presence of Tyrosine- and Tryptophan-coated gold nanoparticles



Effect of reversing the orientation of tyrosine attached to nanoparticles

Because the bioinformatics analysis and tyrosine-insulin docking studies indicate involvement of tyrosine's aromatic moiety, C=O and -NH2 functional groups, the question of whether the orientation of the attached tyrosine molecule is vital to its inhibition effect is very significant. To clarify which functional groups are important for the observed inhibition effect, we strategically (Selvakannan et al. 2013) synthesized AgNPs^{Tyr} sample, where the orientation of the attached tyrosine molecules is reversed (Fig. 1a, c and Figure S2 of supplementary information). The reverse orientation of oxidized tyrosine has been confirmed in our previous study by XPS and zeta potential measurements, wherein the amine group of tyrosine molecule binds with gold nanoparticle surface and the semiquinone group of tyrosine binds with the silver nanoparticle surface (Selvakannan et al. 2013). Hence unlike the gold nanoparticles (AuNPs^{Tyr}), the -NH₂ and C=O groups of the tyrosine molecules in AgNPs^{Tyr} would be protruding outward facing the solvent. Inhibition of amyloid aggregation of insulin was also observed in the presence of such AgNPsTyr nanoparticles. However, the inhibition effect of AgNPs^{Tyr} was found to be slightly lesser than the observed inhibition effect for AuNPs^{Tyr} (Fig. 3d). This data is consistent with the results obtained from insulin-tyrosine docking studies which also suggest the occurrence of interactions mediated through C=O and -NH₂ groups of tyrosine.

Discussion

Current study clearly reveals the potential of the nanoparticles coated with aromatic residues to inhibit amyloid aggregation of insulin. A schematic representation of the current results is given in Fig. 5. Because such inhibition effect is not observed in the presence of isolated tyrosine and tryptophan residues, it is predicted that surface functionalization of these

aromatic residues is critical for their anti-amyloid efficacies. Such functionalization would preferably allow the functional groups of the ligand molecules to participate in crucial interactions with the corresponding functional groups of the protein molecule at the site of binding. Our computational studies predict tyrosine's ability to participate in H-bond, CH- π bond, and electrostatic interactions (Fig. 4a, b). The contribution of the aromatic moiety of tyrosine seems to be larger than the participation mediated through the C=O and -NH₂ groups of tyrosine. This could be one of the reasons behind the slight reduction in the inhibition ability of tyrosine when the molecule's orientation was reversed during surface functionalization (attachment with the nanoparticles through its phenolic –OH group) (Fig. 3e, AgNPs^{Tyr}).

Inhibition of both seed-induced aggregation and spontaneous aggregation reveals two vital clues: (1) These nanoparticles may perhaps bind to insulin monomers undergoing aggregation; (2) They might well bind to the available growing ends of the mature amyloid fibrils blocking the recruitment of monomers. It is also possible that these nanoparticles have the potential to interact with both the monomeric and the fibrillar forms of insulin. This hypothesis is further supported by promotion of the fibril disassembly by AuNPs. Chain B of insulin is known to be the amyloid-prone region of insulin (Tartaglia et al. 2008). Studies have shown that the isolated chain B peptide of bovine insulin also forms amyloid fibrils (Hong and Fink 2005). The site of tyrosine binding, as predicted by our docking studies, falls within the chain B of insulin. Further, the analysis of crystal structures of insulin dimers clearly shows significant contribution by tyrosine residues, particularly at the interface site, connecting two insulin molecules (Fig. 4b, supplemental Information figure S4–S9). Occurrence of such interactions directly supports the possibility of binding of the surface functionalized tyrosine molecules (of the nanoparticles) to insulin by interacting with residues of its chain B and such interactions may interfere with the self-assembly process of insulin monomers into amyloid fibrils. This assumption is further supported by monitoring the change in the conformation of



insulin in an inhibited aggregation reaction which shows the retention of insulin's native conformation (Fig. 3d). However, the $T_{\rm m}$ of insulin in the presence of AuNPs^{Tyr} and AuNPs^{Trp} remained unchanged (Fig. 3a). It is possible that the nanoparticle-insulin interaction may influence the equilibrium of the [Intermediate \(\Lefta \) Aggregate] reaction to shift to the left without enhancing the thermal stability of intermediate species of insulin. This is further supported by the evidence of promotion of insulin fibril dissociation in the presence of tyrosine-coated gold nanoparticles. Aromatic residues are not only important to amyloid aggregation of proteins but they are also known to play critical roles during protein-protein interactions (Bhattacharyya and Chakrabarti 2003). The driving force for amyloid aggregation is believed to arise from favorable interactions between partially folded intermediate species. In the case of globular proteins, such intermediate species are predicted to expose their buried hydrophobic residues to the solvent. Here, the strategy to target intermolecular hydrophobic interaction by nanoparticles coated with aromatic residues is certainly unique and such approach may possibly offer a broader utility of its applications.

In summary, we have identified three important properties of tyrosine- and tryptophan-coated nanoparticles: (a) strong inhibition of spontaneous aggregation of insulin; (b) suppression of seed-induced aggregation of insulin; and (c) promotion of disassembly of insulin amyloid aggregates. Although these results are observed in an in vitro system, it is possible that such inhibition properties may be seen in other amyloid systems including in vivo studies. With such multiple efficacies, these nanoparticles may offer an attractive approach for both production and formulation of effective anti-amyloid candidates.

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Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.

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